



Support of Fuels/Reformer R&D with Computational Fluid Dynamics (CFD) Modeling

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Overall Objective of Reformer Modeling Program

Develop/validate a CFD model of a reformer that can be used to evaluate the interaction between and the impact of fuels (composition) and design/operating parameters on reformer performance.



What does a Computational Fluid Dynamics (CFD) Model Do?

- Solves the laws of conservation of mass, momentum, energy, and atomic species within a flow domain
- *Couples* the conservation laws with other physical & chemical processes (reaction, vaporization, turbulence, etc.) to provide a characterization of major processes within the flow domain
- Computes primary variables throughout the domain: $T, P, \mathbf{U}, \rho, h, k, \varepsilon, Y_i$



Technical Approach

- Develop a reformer model utilizing ANL's multiphase reacting flow CFD codes
 - ◆ Incorporate kinetic models for POX, shift, and PROX reactions
- Model selected micro reactors and prototype reformers
 - ◆ Use CFD models to extract local kinetic rate constants for postulated reduced reaction models
- Validate reformer model
 - ◆ With available data over wide parameter space
- Use validated model to conduct parametric, sensitivity, optimization and scale up studies
 - ◆ Identify most promising fuel composition and associated optimal operating and design parameters
 - ◆ Quantify impact of key operating and design parameters on reformer performance
 - ◆ Explore design options to maximize performance



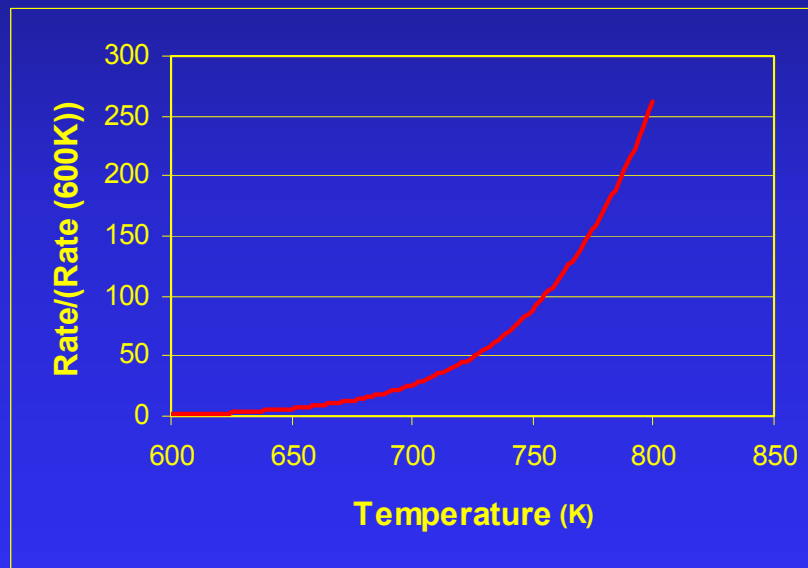
A CFD Model is an Invaluable Tool to Study Fuel/Reformer Interactions

- Fuel constituent reaction rates are characterized by the Arrhenius equation:

$$\frac{\partial C_i}{\partial t} = -K_i \left(\prod_r C_r^{n_r} \right) \exp \left(\frac{-E_i}{RT} \right)$$

Where C_r = concentration of reactants, T = temperature, K_i = preexponential constant, n_r = order of reaction for species r , E_i = activation energy for reaction i

Reaction rates
are very
sensitive to
temperature



**A CFD Model
computes the
local and exit
values of the
flow parameters
& reaction rates
and thus the
overall reformer
performance**

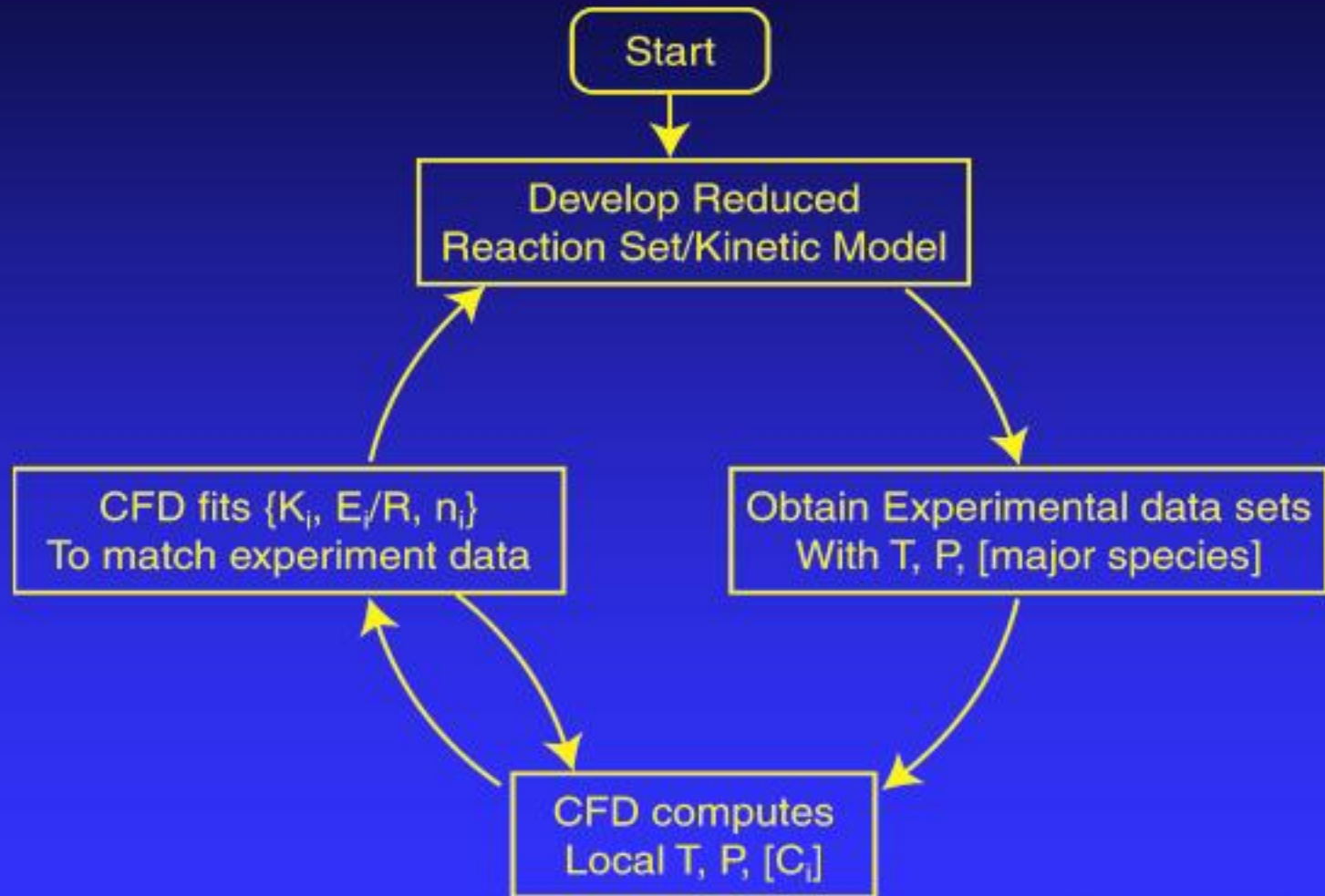


Fuels Composition and Reformer Design Together Determine Overall Reformer Performance

- Critical issues/parameters impacting reformer performance
 - ◆ Operating windows exist with respect to coking limits & catalyst heat tolerance determined by fuel composition, Air/Fuel ratio & Steam/Carbon ratio
 - ◆ Reforming reaction rates of individual fuel constituents
 - ◆ Controlling reforming reactions
 - ◆ Incomplete mixing of reactant streams at catalyst inlet (determined by entry region design)
 - ◆ Heat losses at component boundaries (creates T distribution that impacts reaction rates)
 - ◆ Reformer volume (size)/residence time
- CFD models can be used to help identify optimal fuels/blends and investigate the impact of such issues in a cost effective manner



Kinetic Model Development and Rate Constant Extraction with CFD

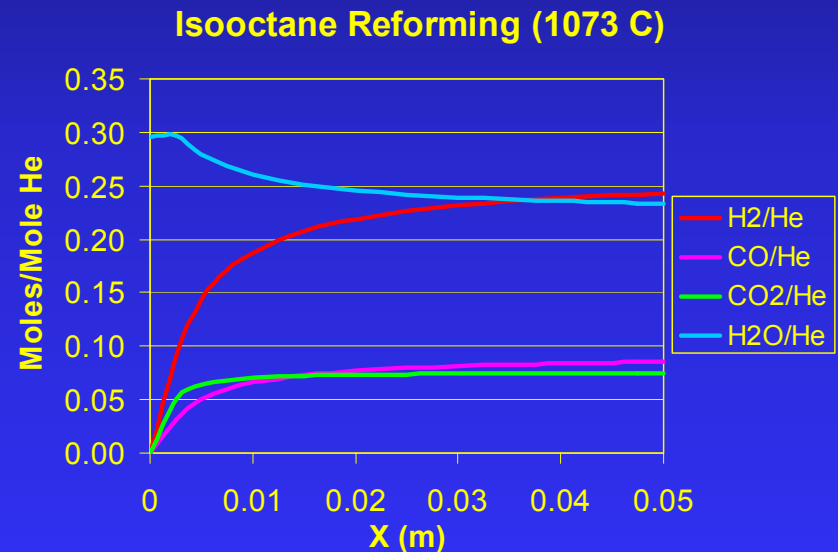


Reduced Isooctane Kinetics Model From Micro Reactor Experiments

■ Reduced reaction set for C_8H_{18} reforming

- ◆ POX: $C_8H_{18} + 4 O_2 \rightarrow 8 CO + 9 H_2$
- ◆ Oxidation: $C_8H_{18} + 25/2 O_2 \rightarrow 8 CO_2 + 9 H_2O$
- ◆ Reforming: $C_8H_{18} + 16 H_2O \rightarrow 8 CO_2 + 25 H_2$
- ◆ Reforming: $C_8H_{18} + 8 H_2O \rightarrow 8 CO + 17 H_2$
- ◆ Shift: $CO + H_2O \rightarrow CO_2 + H_2$

*Typical Predicted
Model Yield Trends in
Micro Reactor*

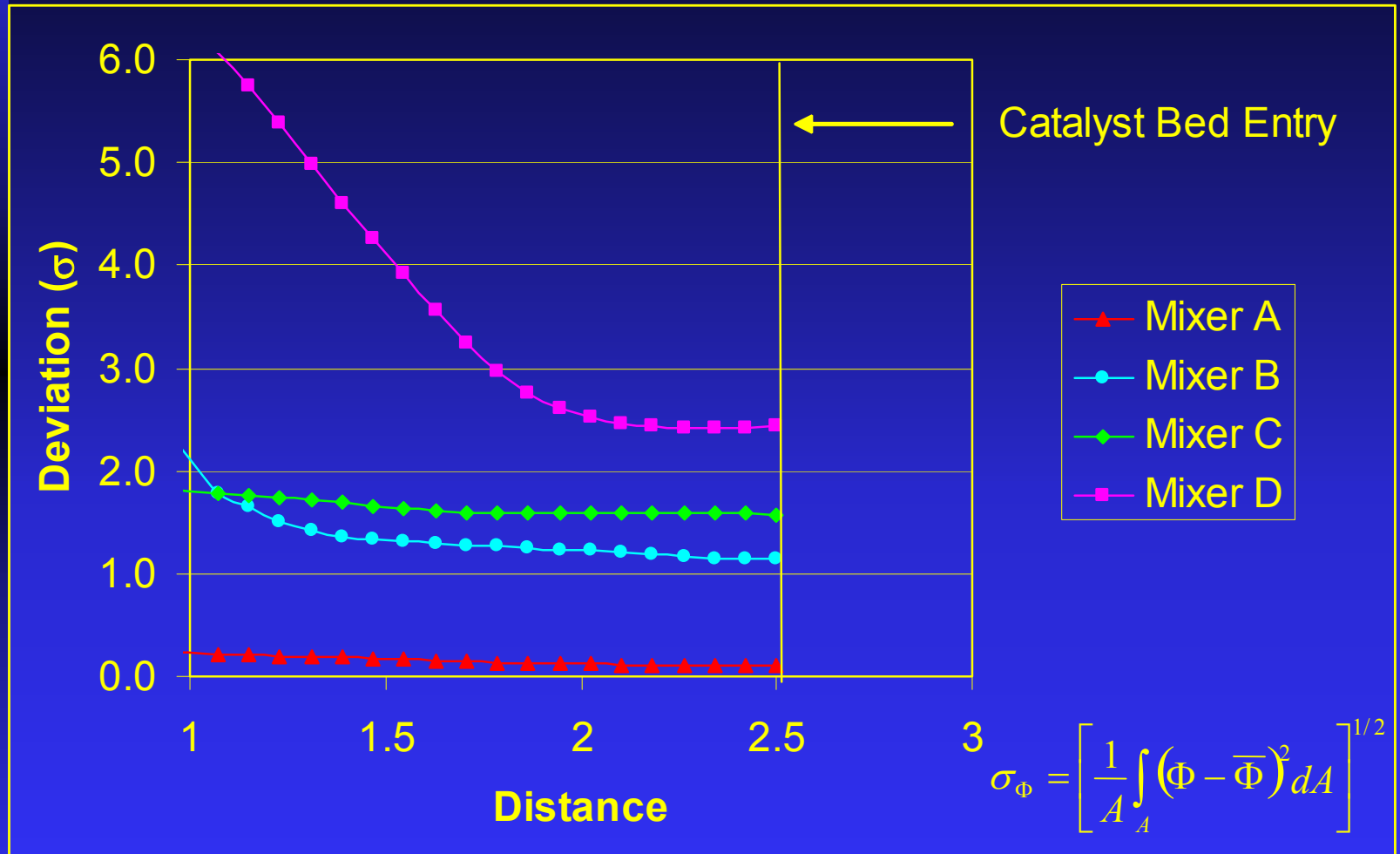




Approach Taken to Evaluate Impact of Reactant Maldistribution at Catalyst Bed Entry on Reformer Performance

- Defined/modeled four different reactant injection designs
 - ◆ Proprietary to industrial partner
 - ◆ Designated as Mixer A, B, C, D
- Coupled entrance models to specified autothermal reformer geometry
- Used isooctane reaction model
- Computed reformer H₂ yields
- Compared results with baseline having uniform reactant distribution at catalyst entry

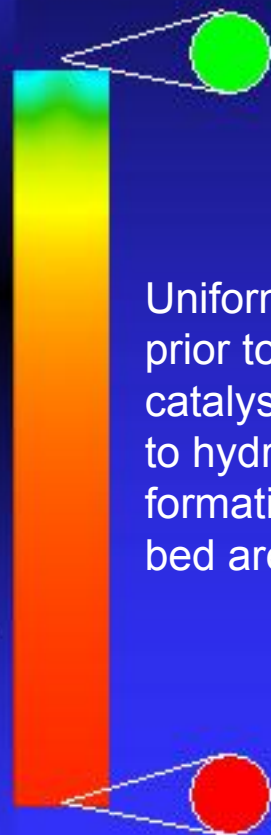
Deviation From Uniform Equivalence Ratio Over Axial Cross Sections in Mixers up to Catalyst Bed Entry





Effect of Incomplete Mixing on Hydrogen Formation in an ATR Catalyst Bed (Based on Mixer D)

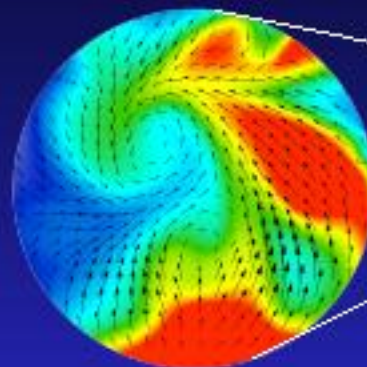
Micro Reformer



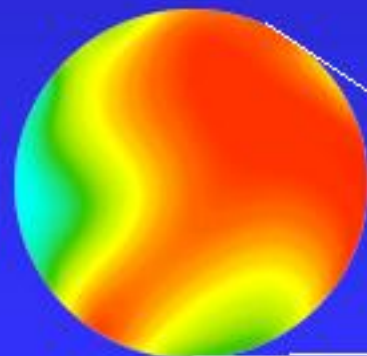
Fuel/Air
mixture ratio
at start of
catalyst bed

Uniform mixing
prior to entering
catalyst bed leads
to hydrogen
formation in entire
bed area

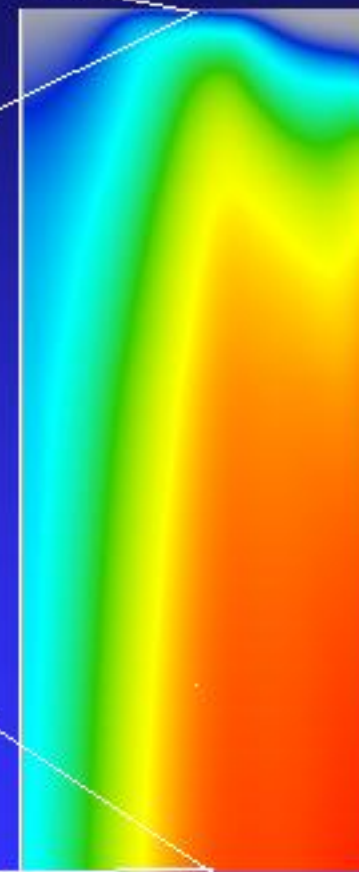
Hydrogen
concentration



Non-uniform mixing
prior to entering
catalyst bed leads to
uneven hydrogen
formation



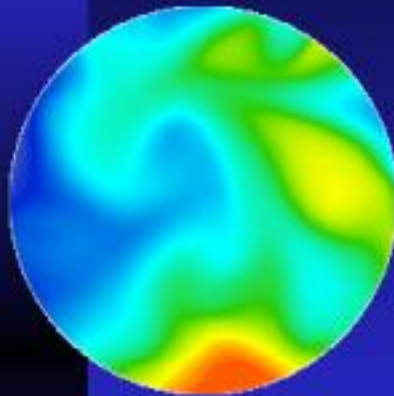
5 kW Reformer



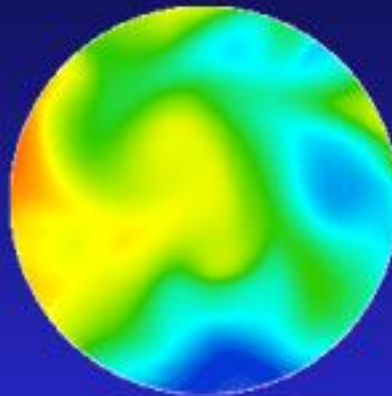


Distributions Traveling Through the Catalyst Bed for Mixer D

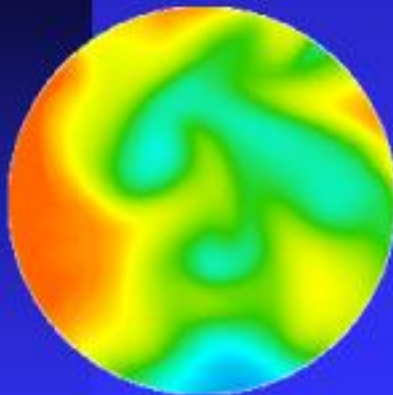
Fuel Concentration



Temperature



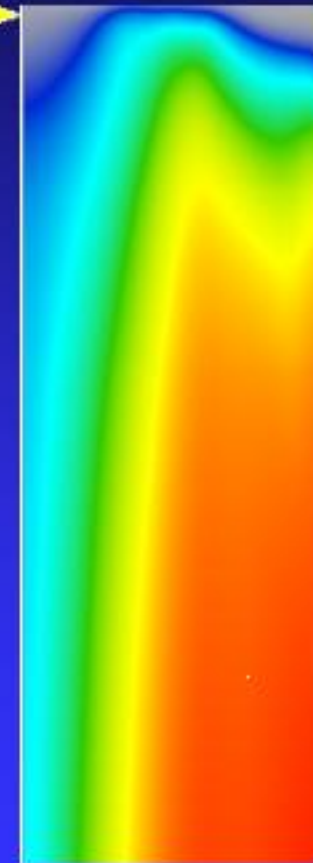
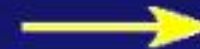
O₂ Concentration



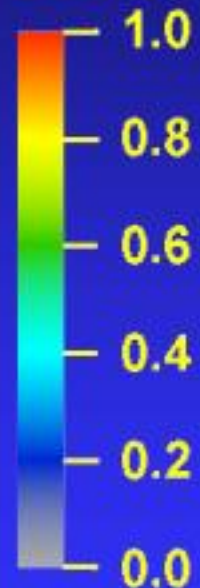
H₂ Concentration



5 kW Reformer

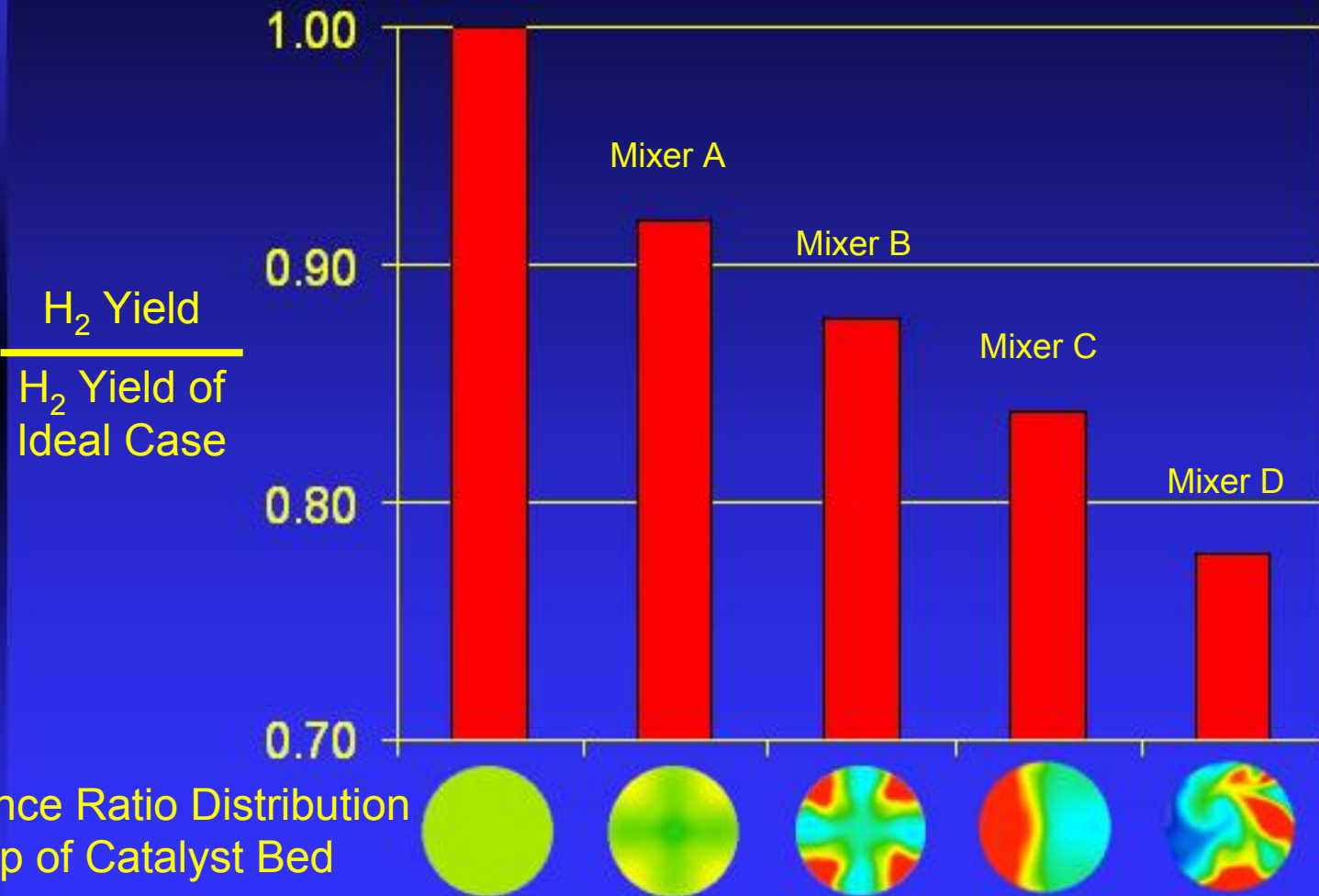


H₂ Conc.





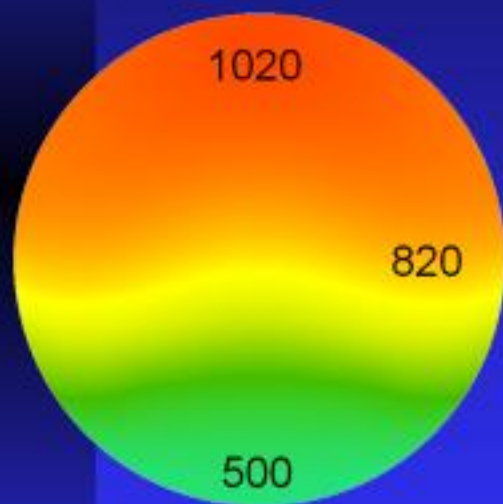
Hydrogen Yield Ratios Compared to Ideal Mixing Case



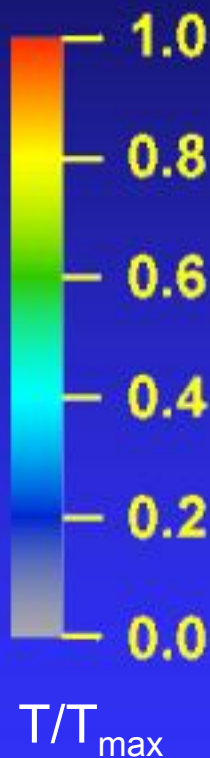


Experimental Thermal Couple Well Temperatures Compared to Computed Trends

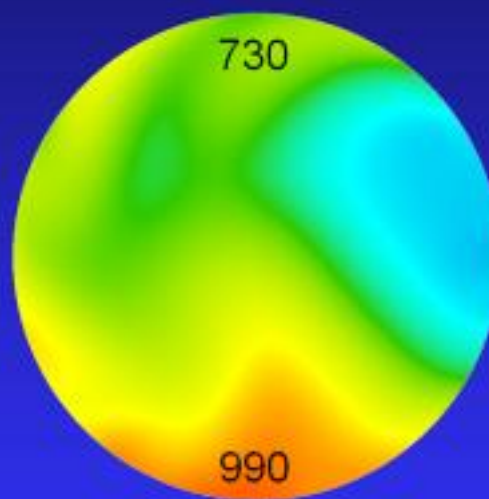
Position at 1/5
Catalyst Bed Length



Mixer C



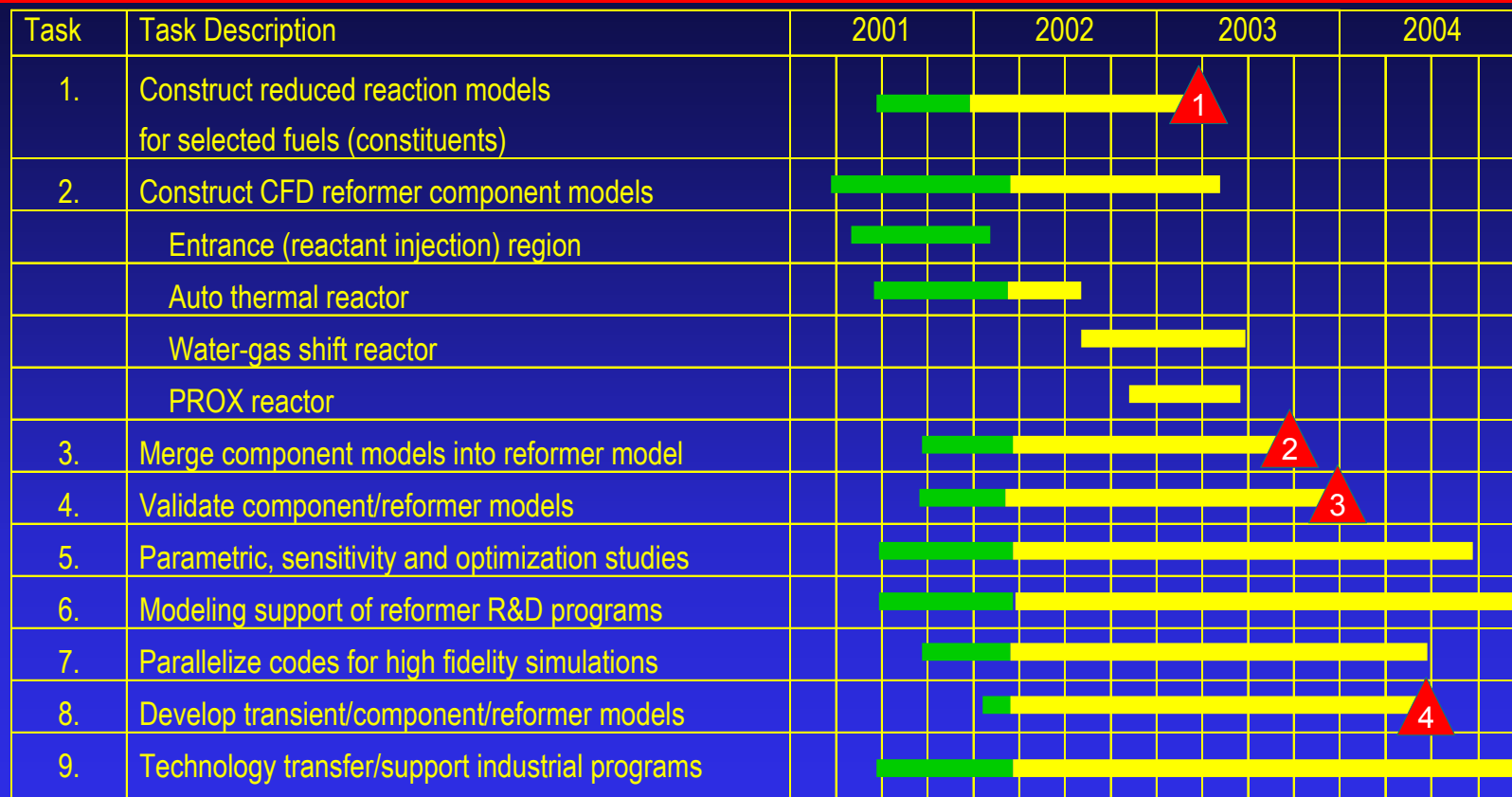
Position at 1/5
Catalyst Bed Length



Mixer D



Program Schedule/Milestones



Major milestones:

Completed work:

1. Reduced reaction models for gasoline fuel developed
3. Component/reformer models validated

2. Overall reformer model completed
4. Transient reformer model completed



Planned Activities for Next Period

- Develop reduced reaction models for selected fuels
 - ◆ Focus on gasoline
 - Start from isooctane model reduced reaction set
 - Identify/use data available from labs and industry
- Finish monolith catalyst model
 - ◆ Add reaction to flow and heat transfer model
 - ◆ Fully couple to mixing zone computation
- Continue transient fast start heat up model development for ATR and water-gas-shift reactor
- Continue ongoing modeling support and CFD analysis activities consistent with guidance and available resources